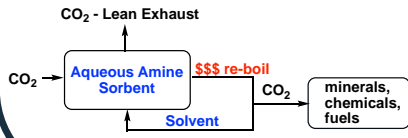


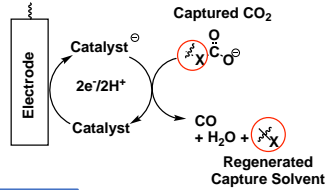
Motivation

CO₂ capture and utilization (CCU) is a vital tool in the fight against climate change
 Mature technologies utilize aqueous scrubbers that are costly to recycle
 Organic liquids with high CO₂ binding affinity have lower energy cost
In situ CO₂ reduction to CO in capture solvent generates a valuable platform chemical via a low emission pathway

Aqueous Amine Scrubbing



Electrocatalytic CCU



Materials And Methods

Electrocatalysis uses an applied potential to generate the energetic driving force for the reaction

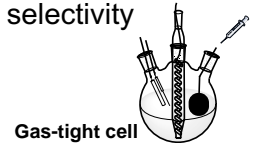


Catalysts, like this iron porphyrin, enable CO₂ reduction to CO over and over without getting used up during the reaction

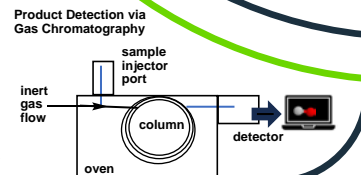
Cyclic Voltammetry (CV)— measures electron transfer events across a range of potentials on a fast time scale

Controlled Potential Electrolysis (CPE)— batch reaction at single potential for determination of products and selectivity

Activity reported as catalyst turnover numbers (TON)



equipped with working, counter, and reference electrodes



Research Highlights and Impact

Carbon Capture and Utilization (CCU) is a strategy to help mitigate CO₂ emissions from industrial post-combustion flue gas

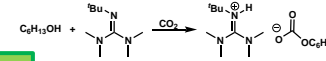


Cleaner emissions
CO

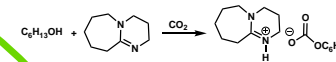


Electrochemical CO₂ reduction to CO via CCU is a green approach to the synthesis of platform chemicals used in detergents, fragrances, pharmaceuticals and more

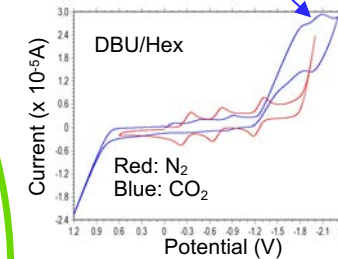
tBuTMG/Hexanol



DBU/Hexanol

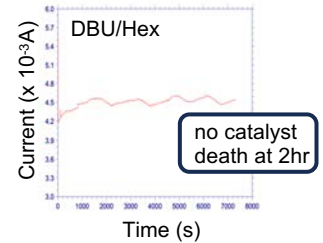


CV Result:



Base/Alcohol mixtures uptake 130 wt % CO₂ and release upon bubbling with N₂

CPE Results*:
CO₂ Reduction at -1.9V vs Ag⁰/+



tBuTMG/Hex: 7 TON CO
 DBU/Hex: 8 TON CO
 *2 hr rxn time

Conclusion & References

Organic solvent mixtures containing both basic and alcohol constituent groups have a **high affinity for CO₂ uptake** while maintaining **low-energy recyclability pathways**

Molecular iron catalysts can **electrochemically transform captured CO₂ to CO directly in the capture solvent**

While proof of concept has been demonstrated, **further studies are needed to optimize the catalytic activity and maximize CO₂ conversion under capture conditions**

(1) Heldebrant, D. J.; Yonker, C. R.; Jessop, P. G.; Phan, L. *Energy & Environmental Science* **2008**. (2) Bhugun, I.; Lexa, D.; Savéant, J.-M. *J. Am. Chem. Soc.* **1996**, *118*, 1769-1776. (3) Siegel, R. E.; Pattanayak, S.; Berben, L. A. *ACS Catalysis* **2022**, 766-784.